

## Microscopic state

In the microscopic consideration, knowledge regarding the constituent particles of the system is taken into account.

- # Suppose, a state of the system is defined by the position coordinates ' $q$ ' and momentum coordinates ' $p$ '.
- # For a system of  $N$  particles,  
 $q$  represents a set of  $3N$  position ~~coordinates~~ coordinates,  $q_1, q_2, \dots, q_{3N}$   
 $p$  represents a set of  $3N$  momentum coordinates  $P_1, P_2, \dots, P_{3N}$ .  
 (It has taken into account all the three axes  $x, y, z$ . That's why it is shown in  $3N$  ways).

For one set of ~~of~~ position and momentum coordinates, the system is represented by one point in the phase ~~of~~ space. Such state of the system is known as a microscopic state or microstate.

THAT'S

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Thus, from the microscopic point of view, the macroscopic description of the system is not applicable.

Therefore, a microstate of a macroscopically defined system is described through a probability density  $f(q, p, t)$  which is a function of position and momentum coordinates of the constituent particles and time  $t$ .